

Chapter 7:
Hemoglobin:
Portrait of protein
in action

Chapter 4

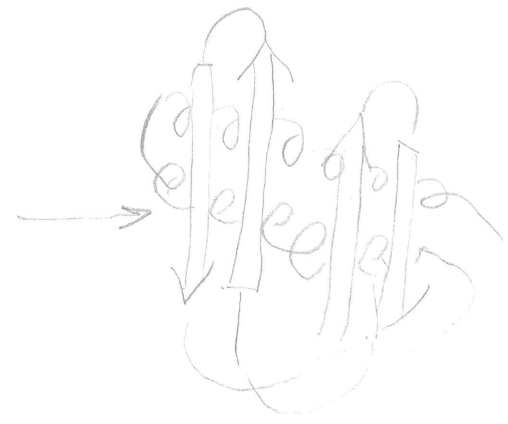
3D structure of proteins

* Structure-function relationship of protein

prion protein: mad cow Disease
Spongiform encephalopathy
Kuru - Creutzfeldt - Jakob Disease



PrP^C
cellular form



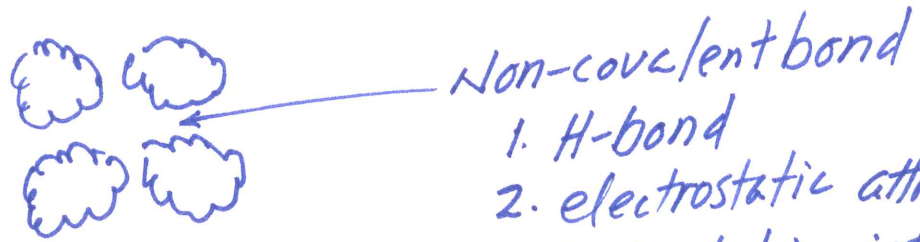
PrP^{Sc}
scrapie form
infectious
cross species

* primary structure: amino acid sequence

* Secondary structure: α -helix
 β -sheet
domains (supersecondary structure)

Tertiary Structure: side-chain + pos prosthetic group

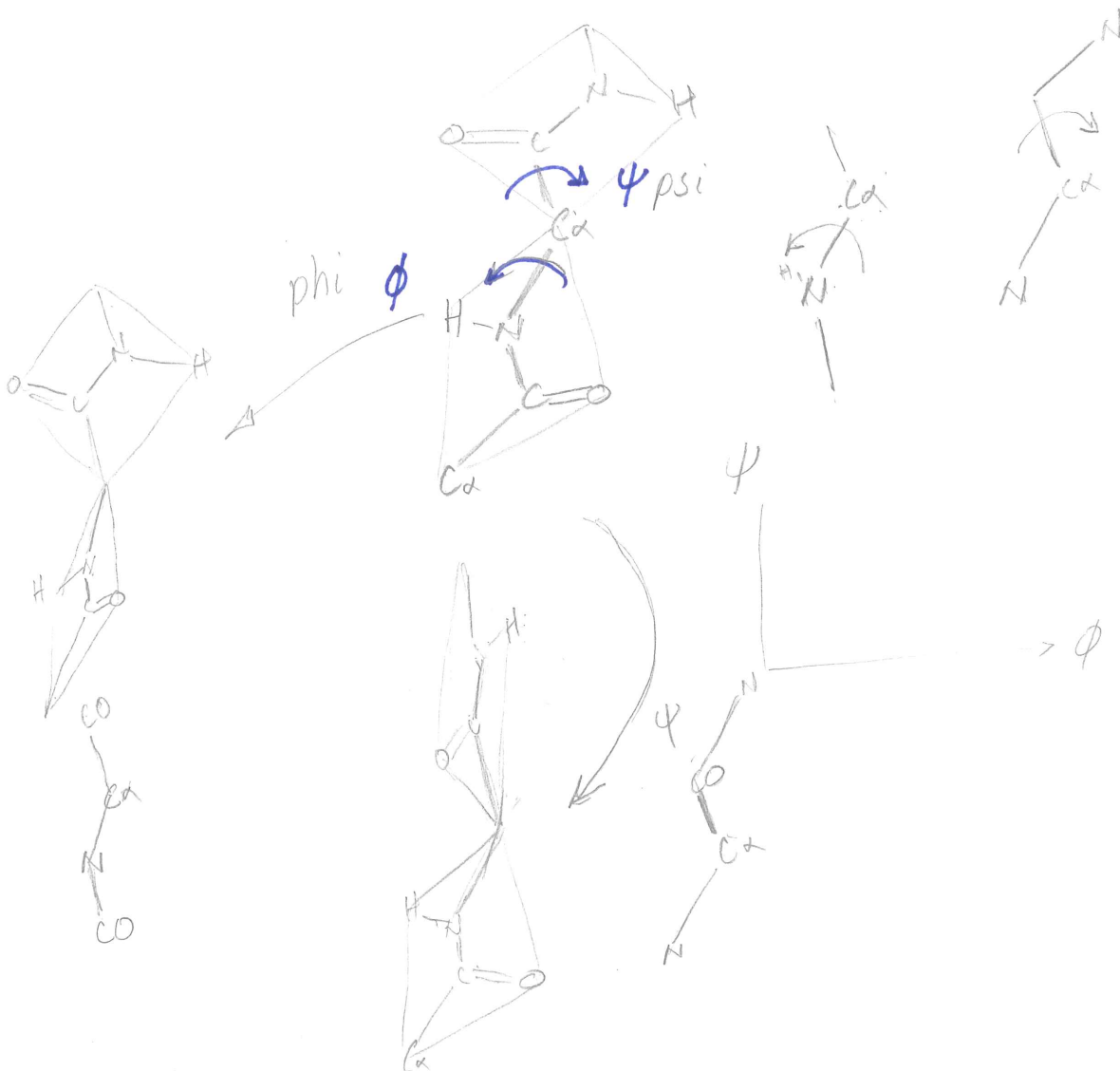
Quaternary Structure: multi-subunit protein



1. H-bond
2. electrostatic attraction
3. hydrophobic interaction

* Secondary structure

periodic structure in protein backbone



protein 3D structure Determination

X-ray crystallography
technique

Saturate solution

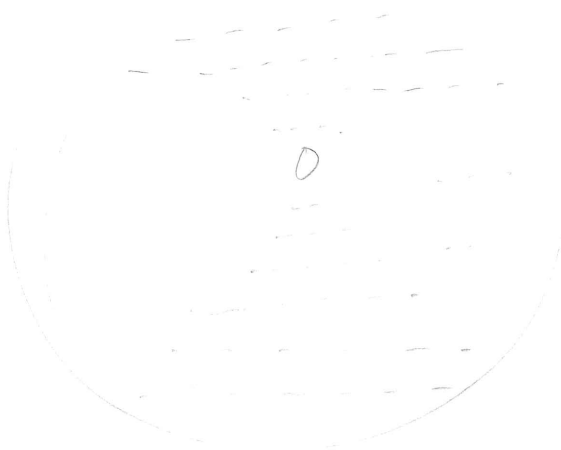
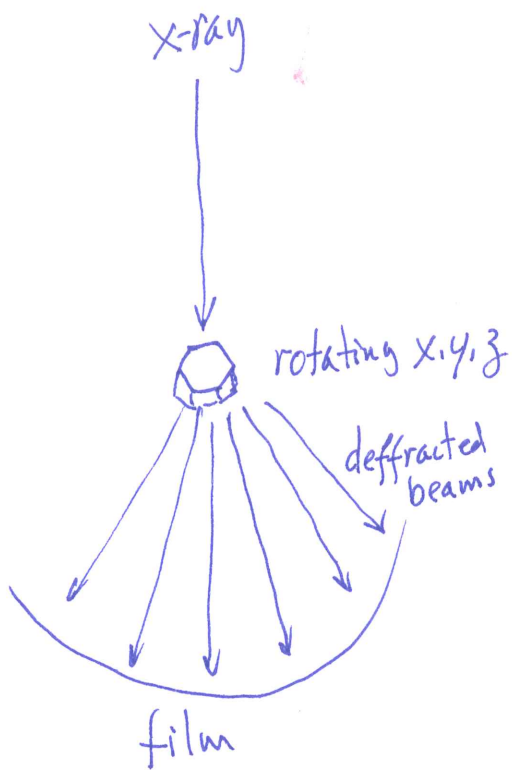
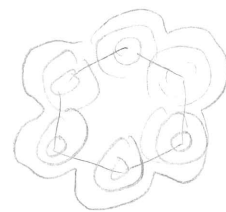
↓
single crystal

↓
diffractometer 繞射

↓
diffraction pattern
Fourier Transform (Image construction)

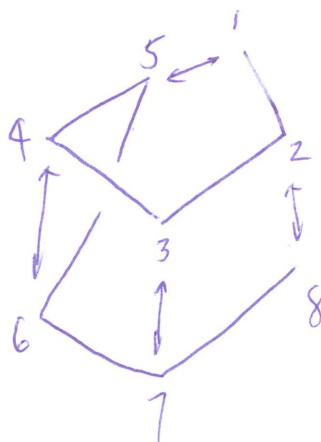
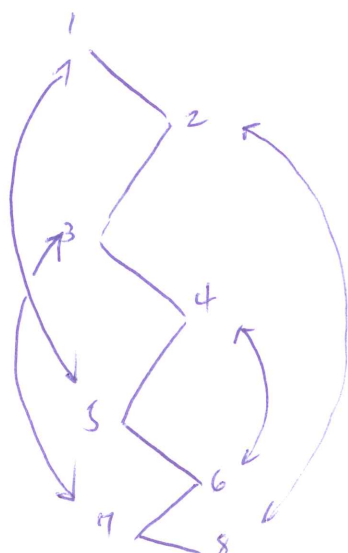
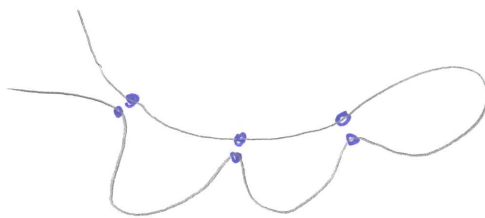
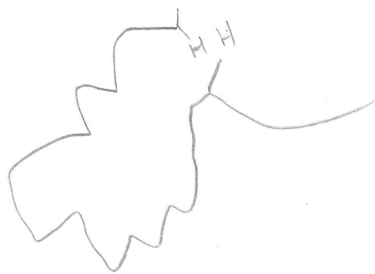
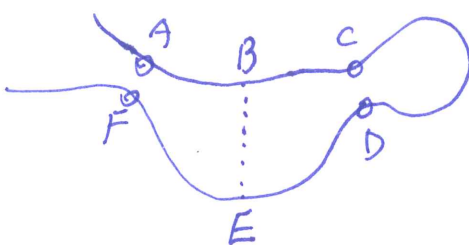
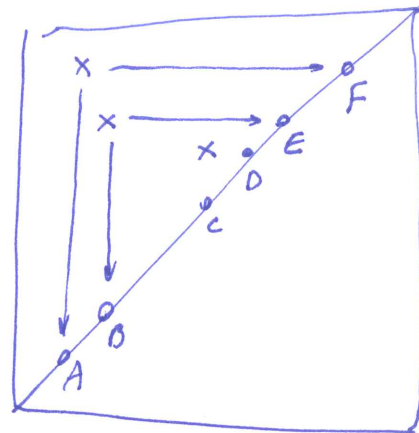
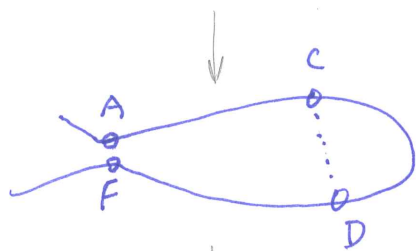
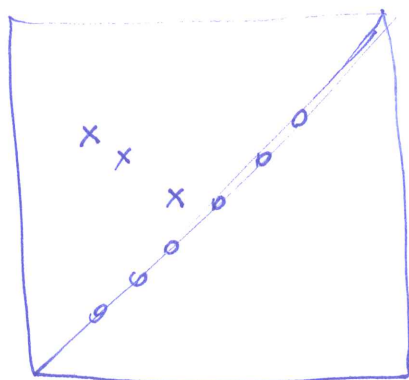
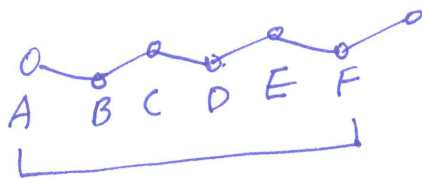
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Electron density

↓
3D structure



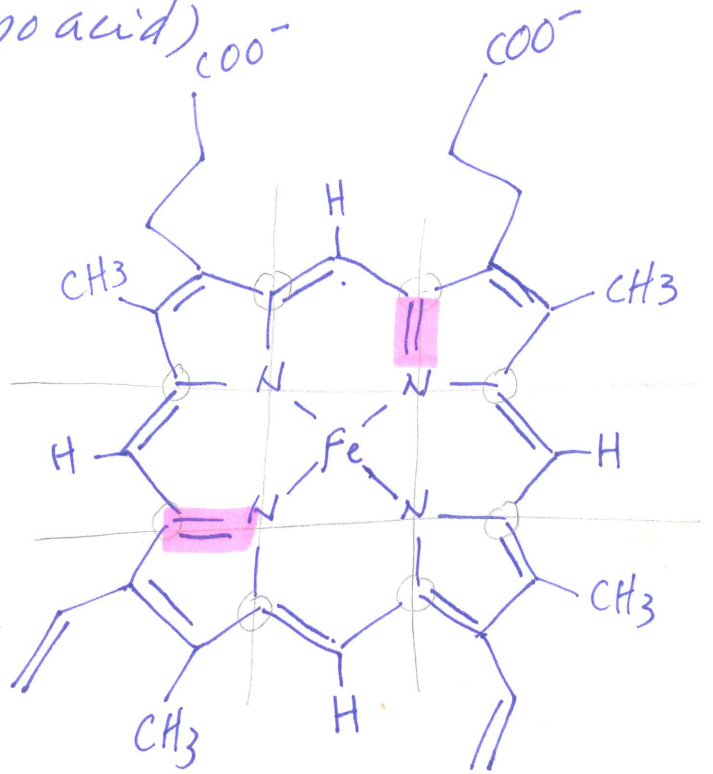
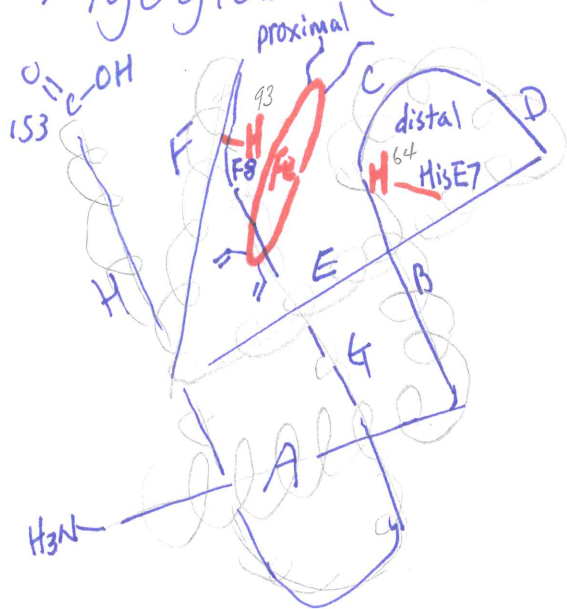
diffraction pattern
in film

NMR Spectroscopy : NOEs (Nucleus Overhauser Effect)

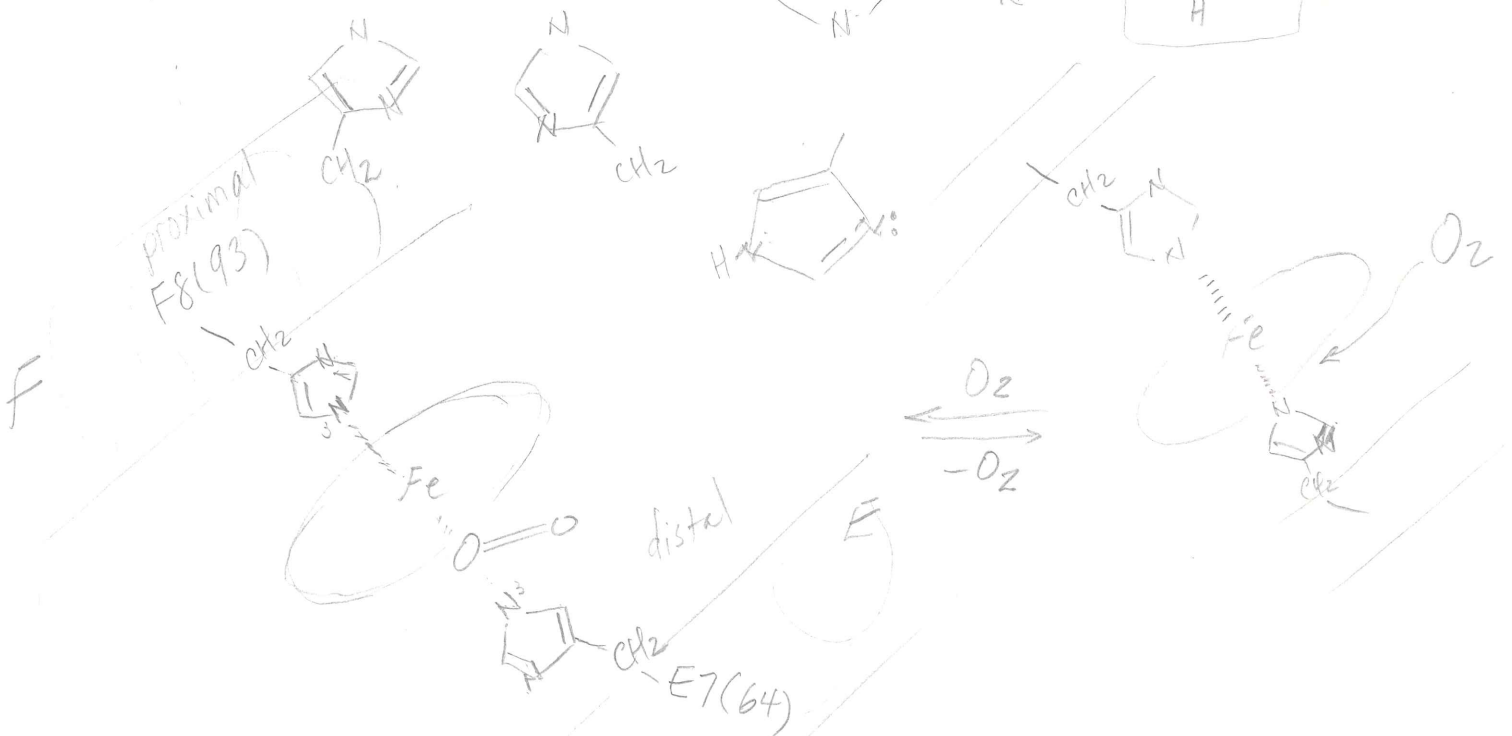
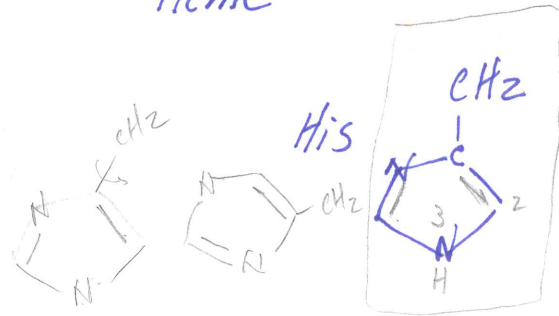
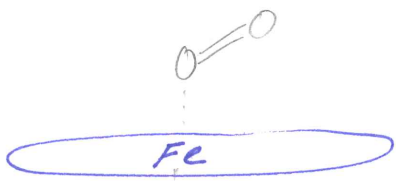


Hemoglobin

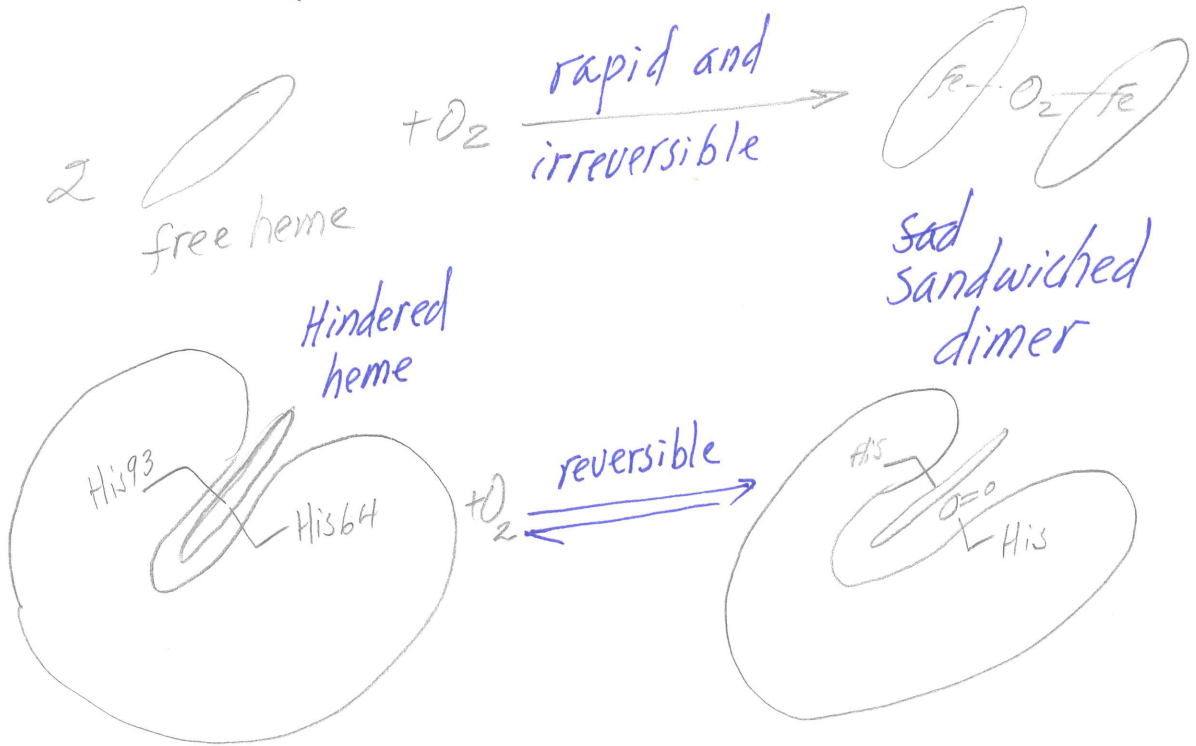
Myoglobin (153 amino acid) COO^-



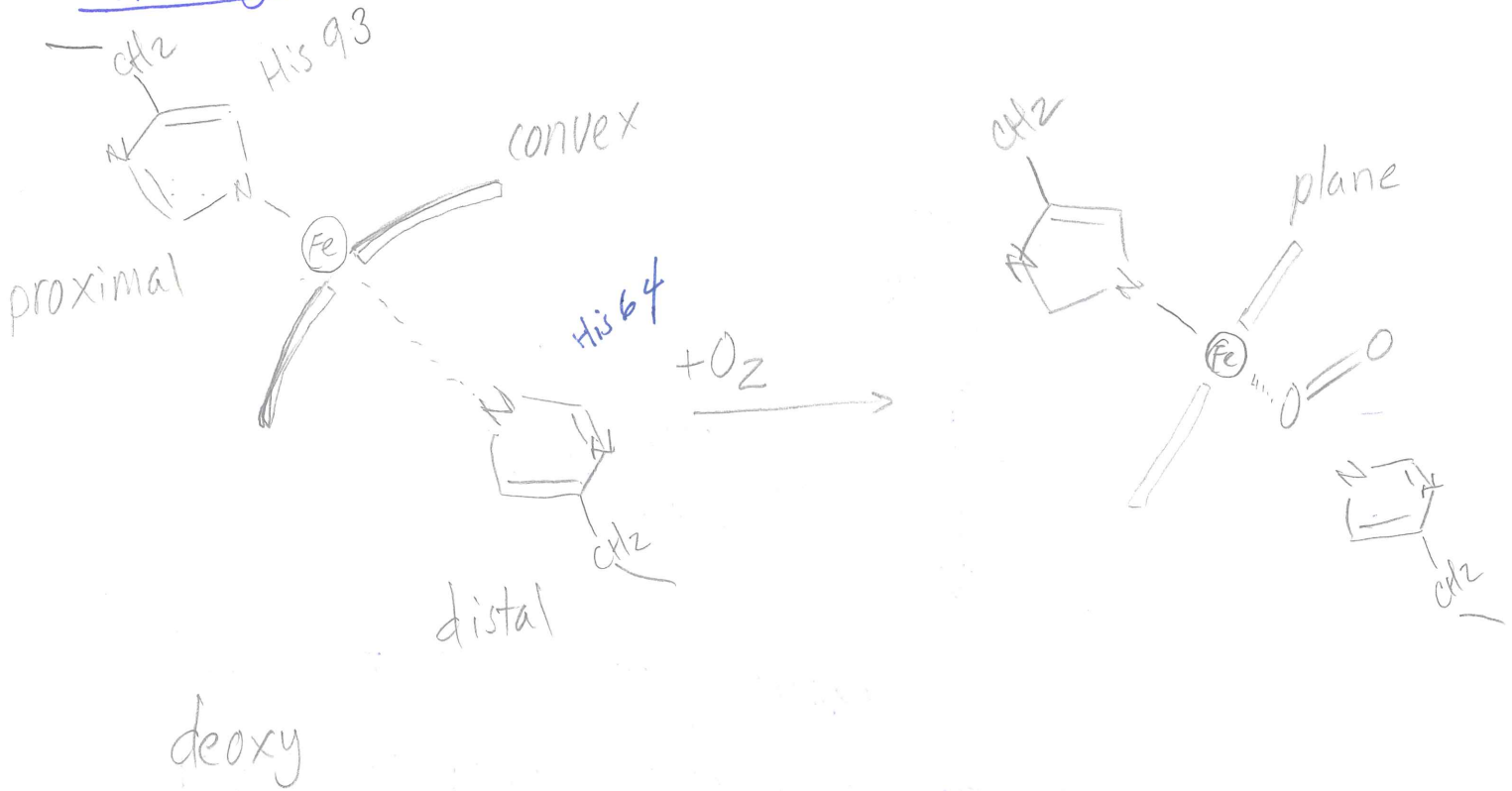
Heme

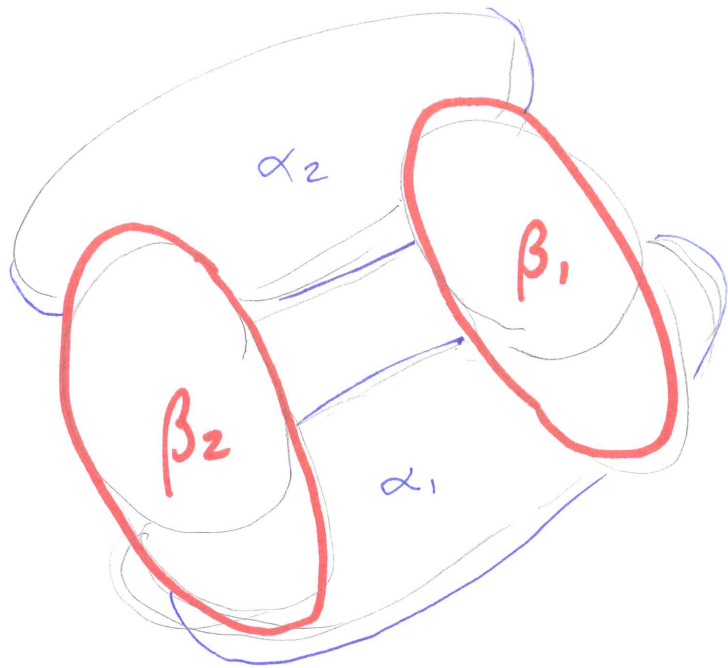


Hindered Heme environment is essential for O_2 transport and storage

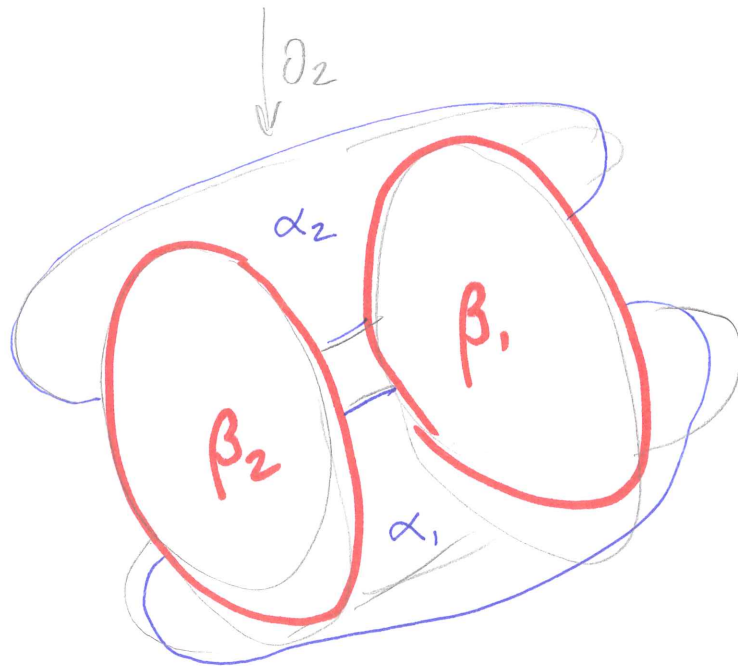


Hemoglobin



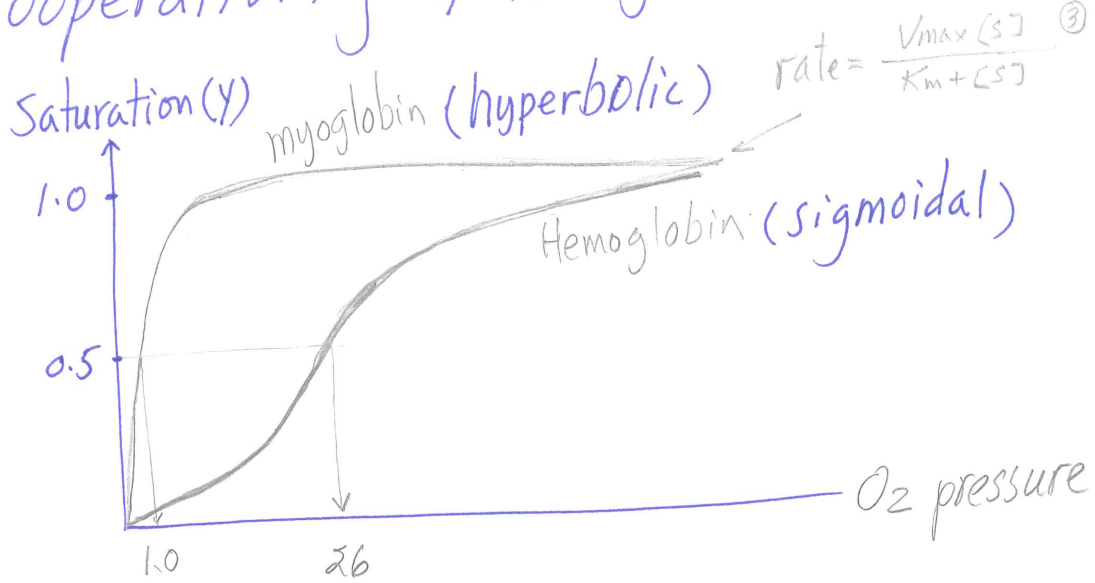


de-oxy
Hemoglobin



Oxy-
Hemoglobin

Cooperativity of Hemoglobin



- ① $[S] \ll K_m$
- ② $[S] \gg K_m$
- ③ $[S] = K_m$

Oxygen dissociation curve

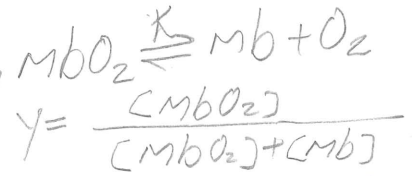
① hyperbolic / sigmoidal

② Myoglobin > Hemoglobin

③ Myoglobin: $Y = \frac{PO_2}{PO_2 + P_{50}}$

Hemoglobin:

$$Y = \frac{(PO_2)^n}{(PO_2)^n + (P_{50})^n} \quad \text{Hill}$$

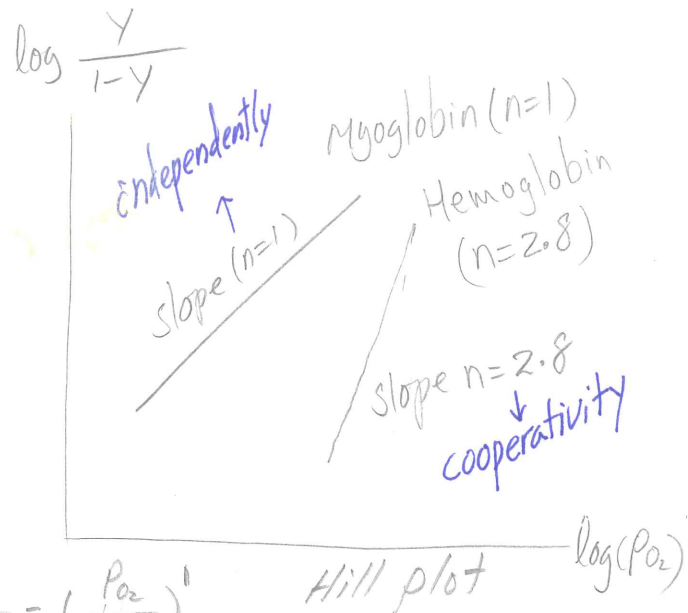
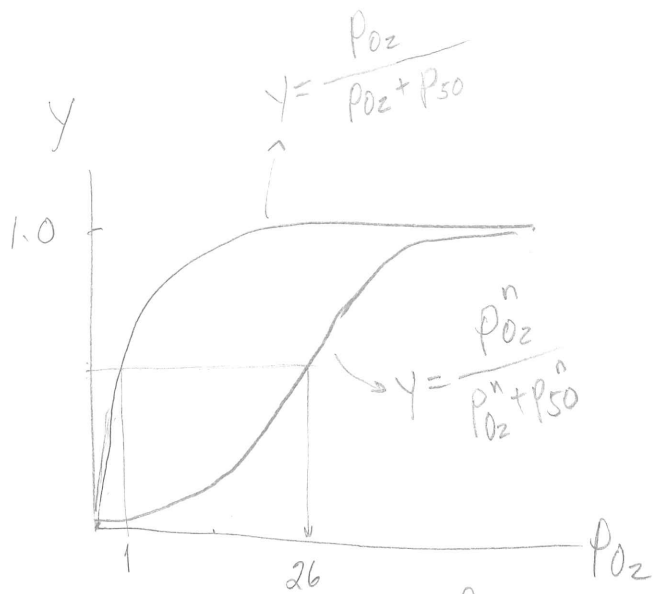


Michaelis-Menten

$$K = \frac{[Mb][O_2]}{[MbO_2]} ; Y = \frac{[MbO_2]}{[MbO_2] + [Mb]}$$

$$\left. \begin{aligned} [O_2] &= K \frac{[MbO_2]}{[Mb]} \\ [MbO_2] &= \frac{[O_2][Mb]}{K} \end{aligned} \right\} Y = \frac{\frac{[O_2][Mb]}{K}}{\frac{[O_2][Mb]}{K} + [Mb]} = \frac{[O_2]}{[O_2] + K}$$

$$Y = \frac{[O_2]}{[O_2] + K} = \frac{PO_2}{PO_2 + P_{50}}$$

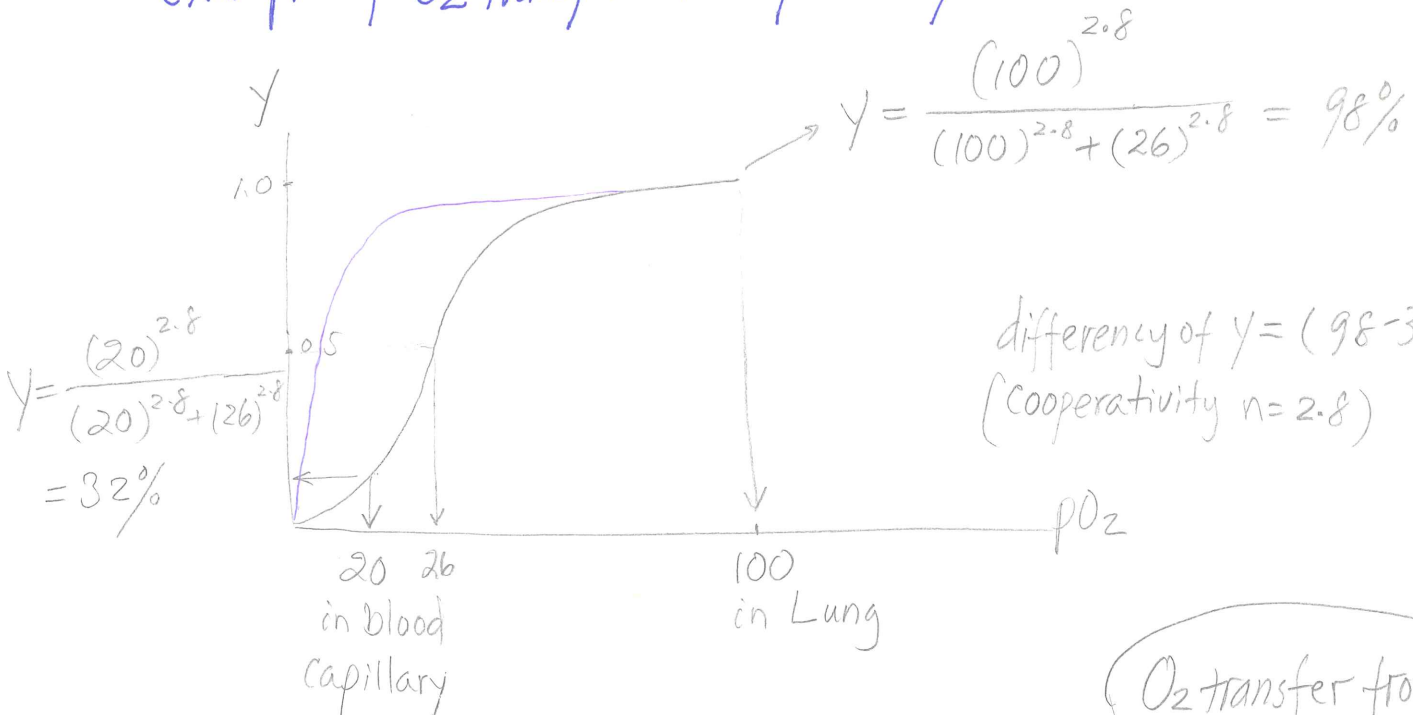


Myoglobin: $y = \frac{PO_2}{PO_2 + P_{50}} \Rightarrow \frac{y}{1-y} = \left(\frac{PO_2}{P_{50}}\right)^1$

Hemoglobin: $y = \frac{PO_2^n}{PO_2^n + P_{50}^n} \Rightarrow \frac{y}{1-y} = \left(\frac{PO_2}{P_{50}}\right)^n$

$\log \frac{y}{1-y} = n \log \left(\frac{PO_2}{P_{50}}\right)$

Example of O₂ transfer w/ cooperativity



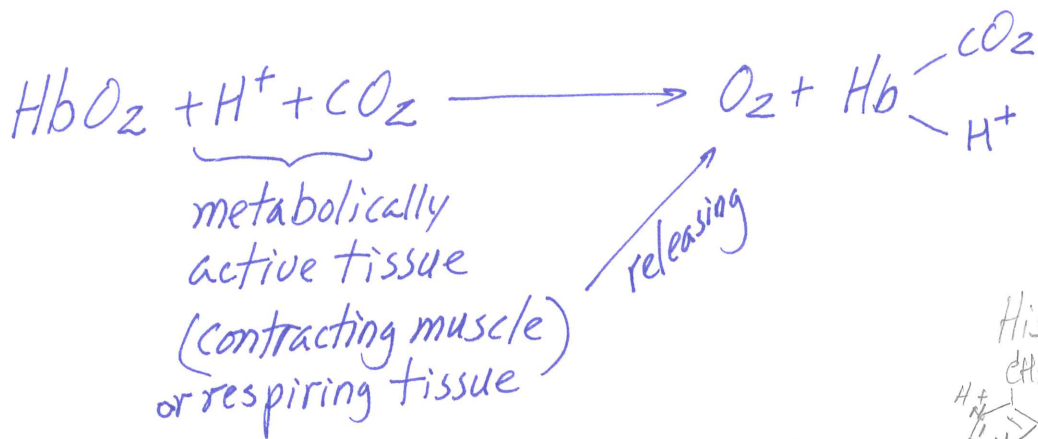
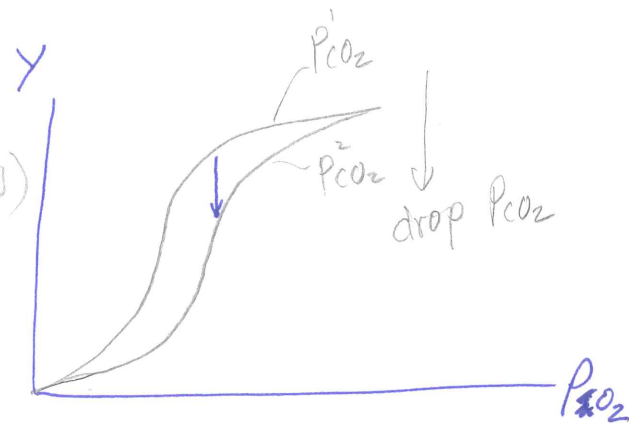
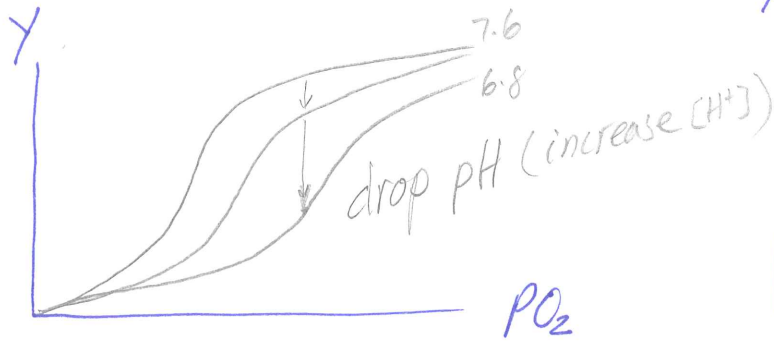
differency of $y = (98 - 32)\% = 0.66$
(cooperativity $n = 2.8$)

If non-cooperativity ($n = 1$)

Lung: $y = \frac{100}{100 + 26} = 0.79$
 blood: $y = \frac{20}{20 + 26} = 0.43$
 } (driving force) difference = 0.38

O₂ transfer from Lung → blood
 differences $\frac{0.66}{0.38} = 1.8$
 different

Bohr effect (H^+ & CO_2)



BPG (2,3-bisphosphoglycerate) binding

